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Key indicators

Single-crystal X-ray study

T = 293 K

Mean $\sigma(\text{C}-\text{C}) = 0.003 \text{ \AA}$

R factor = 0.030

wR factor = 0.086

Data-to-parameter ratio = 8.0

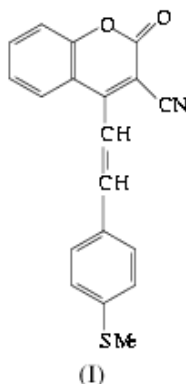
For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.

3-Cyano-4-[2-(4-methylthiophenyl)ethenyl]-2H-1-benzopyran-2-one

In the title compound, 4-[2-(4-methylthiophenyl)ethenyl]-2-oxo-2H-1-benzopyran-3-carbonitrile, $\text{C}_{19}\text{H}_{13}\text{NO}_2\text{S}$, the benzopyran and phenyl rings are individually planar, but the phenyl ring is twisted $56.7 (1)^\circ$ out of the benzopyran ring plane. The configuration about the ethenyl double bond is *E*.

Comment

The structure determination of the title compound, (I), was taken up as part of our studies on coumarin derivatives which possess a variety of medicinal and biological properties (Parrish *et al.*, 1974; Evans *et al.*, 1981; Fujiwara *et al.*, 1978; Song & Gordon, 1970; Kawase *et al.*, 2001). It is of physiological interest that there is an apparent close chemical similarity between coumarin and vitamin K (Kralt & Claassen, 1972).



The molecular geometry of (I) is similar to that of 3-cyano-6-methyl-4-[2-(4-methoxyphenyl)ethenyl]-2H-1-benzopyran-2-one (Vijayalakshmi *et al.*, 2001). The bond lengths and angles in the coumarin moiety agree well with the reported values (Jha *et al.*, 2000; Chinnakali *et al.*, 1998, 1999; Vijayalakshmi *et al.*, 2000, 2001). The dihedral angle between the phenyl and planar benzopyran rings is $56.7 (1)^\circ$. The widening of the bond angle C11–C12–C18 to $126.8 (2)^\circ$ is due to the close approach of the H11 and H17 atoms (2.256 Å). A similar feature is observed in the structures of cinnamanilides (Renganayaki *et al.*, 1999, 2000; Subramanian *et al.*, 1999) and dienethioamide (Nesterov *et al.*, 2000). Also, the slight increase in C4–C11–C12 to $122.2 (2)^\circ$ is due to steric repulsion between H12 and C21 (H12...C21 2.629 Å). The $\text{C}_{\text{sp}^2}-\text{S}$ [1.755 (2) Å] and $\text{C}_{\text{sp}^3}-\text{S}$ [1.777 (4) Å] distances show partial double-bond character (Malhotra *et al.*, 1997; Azim *et al.*, 1997; Kumar *et al.*, 1999; Allen *et al.*, 1987). A C–H...N intermolecular short contact is observed

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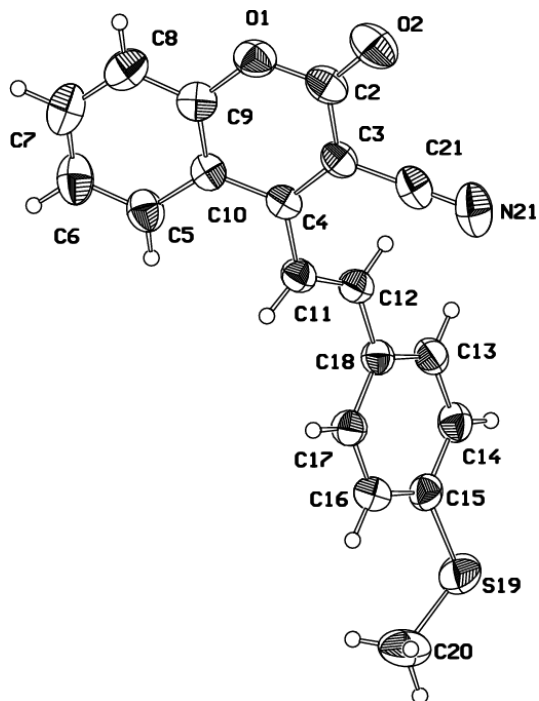


Figure 1
The molecular structure of (I) showing 50% probability displacement ellipsoids.

[C16...N21ⁱ 3.337 (4) Å and H16...N21ⁱ 2.56 Å; symmetry code: (i) $x, y + 1, z$].

Experimental

A mixture of 3-cyano-4-methyl-2*H*-1-benzopyran-2-one (0.01 mol) and 4-methylthiobenzaldehyde (0.01 mol) was dissolved in chloroform (75 ml) and a few drops of piperidine (8–10 drops) was added as catalyst. The mixture was heated on a hotplate with stirring for 15–16 h. After evaporation of the solvent, the solid residue was recrystallized from dimethylformamide to give yellow crystals (m.p. 482 K; yield 64%).

Crystal data

C₁₉H₁₃NO₂S
 $M_r = 319.36$
 Orthorhombic, $P2_12_12_1$
 $a = 7.783$ (3) Å
 $b = 7.8653$ (10) Å
 $c = 25.610$ (3) Å
 $V = 1567.7$ (7) Å³
 $Z = 4$
 $D_x = 1.353$ Mg m⁻³

Cu $K\alpha$ radiation
 Cell parameters from 25 reflections
 $\theta = 2$ –25°
 $\mu = 1.91$ mm⁻¹
 $T = 293$ (2) K
 Needle, yellow
 0.15 × 0.10 × 0.09 mm

Data collection

Enraf–Nonius CAD-4 diffractometer
 ω –2 θ scans
 Absorption correction: ψ scan (North *et al.*, 1968)
 $T_{\min} = 0.784$, $T_{\max} = 0.891$
 1676 measured reflections
 1676 independent reflections

1606 reflections with $I > 2\sigma(I)$
 $\theta_{\max} = 67.9^\circ$
 $h = 0 \rightarrow 9$
 $k = 0 \rightarrow 9$
 $l = 0 \rightarrow 30$
 3 standard reflections every 100 reflections
 intensity decay: negligible

Refinement

Refinement on F^2
 $R[F^2 > 2\sigma(F^2)] = 0.030$
 $wR(F^2) = 0.086$
 $S = 1.06$
 1676 reflections
 210 parameters
 H-atom parameters constrained

$w = 1/[\sigma^2(F_o^2) + (0.0501P)^2 + 0.3363P]$
 where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{\max} < 0.001$
 $\Delta\rho_{\max} = 0.18$ e Å⁻³
 $\Delta\rho_{\min} = -0.21$ e Å⁻³
 Extinction correction: *SHELXL97*
 Extinction coefficient: 0.0037 (4)

Table 1

Selected torsion angles (°).

C3–C4–C11–C12	–45.5 (3)	C16–C15–S19–C20	4.9 (3)
C10–C4–C11–C12	135.5 (2)	C14–C15–S19–C20	–175.5 (2)

All H atoms were fixed using geometrical considerations. The absolute configuration is indeterminate for the title compound.

Data collection: *CAD-4 Software* (Enraf–Nonius, 1989); cell refinement: *MolEN* (Fair, 1990); data reduction: *MolEN*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *ZORTEP* (Zsolnai, 1997); software used to prepare material for publication: *SHELXL97*.

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References

- Allen, F. H., Kennard, O., Watson, D. G., Brammer, L., Orpen, A. G. & Taylor, R. (1987). *J. Chem. Soc. Perkin Trans. 2*, pp. S1–19.
- Azim, A., Parmar, V. S. & Errington, W. (1997). *Acta Cryst. C53*, 1436–1438.
- Chinnakali, K., Fun, H.-K., Sriraghavan, K. & Ramakrishnan, V. T. (1998). *Acta Cryst. C54*, 367–368.
- Chinnakali, K., Fun, H.-K., Sriraghavan, K. & Ramakrishnan, V. T. (1999). *Acta Cryst. C55*, 946–948.
- Enraf–Nonius (1989). *CAD-4 Software*. Version 5.0. Enraf–Nonius, Delft, The Netherlands.
- Evans, J. M., Showell, G. A. & Fake, C. S. (1981). *Chem. Abstr.* **95**, 115301.
- Fair, C. K. (1990). *MolEN*. Enraf–Nonius, Delft, The Netherlands.
- Fujiwhara, M., Sasaki, T. & Uchida, T. (1978). *Chem. Abstr.* **89**, 14799.
- Jha, A., Malhotra, S., Parmar, V. S. & Errington, W. (2000). *Acta Cryst. C56*, 899–900.
- Kralt, T. & Claassen, V. (1972). *Drug Design*, Vol. III, edited by E. J. Ariens, pp. 189–203. New York: Academic Press.
- Kawase, M., Varu, B., Shah, A., Motohashi, N., Tani, S., Saito, S., Devnath, S., Mahapatra, S. G. & Chakrabarty, A. N. (2001). *Arzneim. Forch. (Drug Res. II)*, **51**, 67–71.
- Kumar, R., Parmar, V. S. & Errington, W. (1999). *Acta Cryst. C55*, 561–563.
- Malhotra, S., Parmar, V. S. & Errington, W. (1997). *Acta Cryst. C53*, 1442–1444.
- Nesterov, V. N., Antipin, M. Yu., Timofeeva, T. V. & Clark, R. D. (2000). *Acta Cryst. C56*, 88–89.
- North, A. C. T., Phillips, D. C. & Mathews, F. S. (1968). *Acta Cryst. A24*, 351–359.
- Parrish, J. A., Fitzpatrick, T. B., Tannenbaum, L. & Pathak, M. A. (1974). *New Engl. J. Med.* **291**, 206–209.
- Renganayaki, S., Subramanian, E., Shanmuga Sundara Raj, S. & Fun, H.-K. (1999). *Acta Cryst. C55*, 1672–1673.
- Renganayaki, S., Subramanian, E., Shanmuga Sundara Raj, S. & Fun, H.-K. (2000). *Acta Cryst. C56*, 349–350.

- Sheldrick, G. M. (1997). *SHELXS97* and *SHELXL97*. University of Göttingen, Germany.
- Song, P. S. & Gordon, W. H. (1970). *J. Phys. Chem.* **74**, 4234–4240.
- Subramanian, E., Renganayaki, S., Shanmuga Sundara Raj, S. & Fun, H.-K. (1999). *Acta Cryst.* **C55**, 764–766.
- Vijayalakshmi, L., Parthasarathi, V., Varu, B., Dodia, N. & Shah, A. (2000). *Acta Cryst.* **C56**, e401–402.
- Vijayalakshmi, L., Parthasarathi, V., Varu, B. & Shah, A. (2001). *Acta Cryst.* **E57**, o245–246.
- Zsolnai, L. (1997). *ZORTEP*. University of Heidelberg, Germany.